

**Series 4**

**03 October 2025**

**Exercise 1 Interstitials: Fe-C**

What is the maximum elastic distortion of the body-centered cubic iron network (iron) from introducing a carbon interstitial (atomic radius of iron = 1.24 Å)? Compare it with the distortion in the face-centered cubic iron (atomic radius of iron is 1.27 Å) and qualitatively deduce the respective carbon solubility in the two phases.

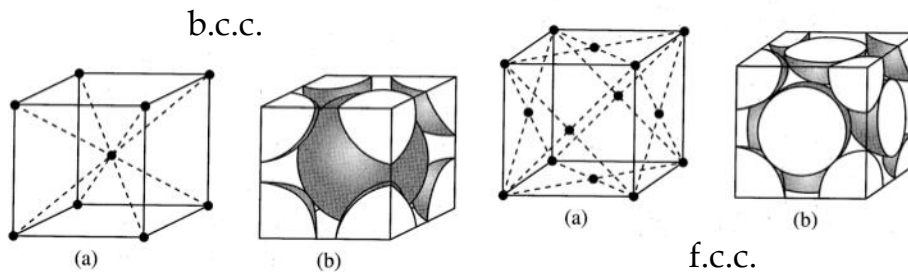


Fig. 4.1 Compact packing

**Exercise 2: Equilibrium concentration of vacancies**

Calculate the equilibrium concentration of vacancies in a metal at constant pressure and temperature.

**Exercise 3: Thermal expansion of aluminum**

We measure the thermal expansion of aluminum by two methods: 1) by measuring the

dilatation strain  $\frac{\Delta L}{L}$  of a sample and 2) by measuring the relative variation of the lattice

parameter  $a$  by X-ray diffraction. The slopes of the curves obtained with these two methods do not coincide (see Figure 4.1). Why?

The difference between these measurements at 400°C is  $\left(\frac{\Delta L}{L} - \frac{\Delta a}{a}\right) = 7.633 \cdot 10^{-6}$  ;

At 600°C, it is  $\left(\frac{\Delta L}{L} - \frac{\Delta a}{a}\right) = 2.94 \cdot 10^{-4}$  .

Knowing these data, calculate the formation energy of vacancies in aluminum.

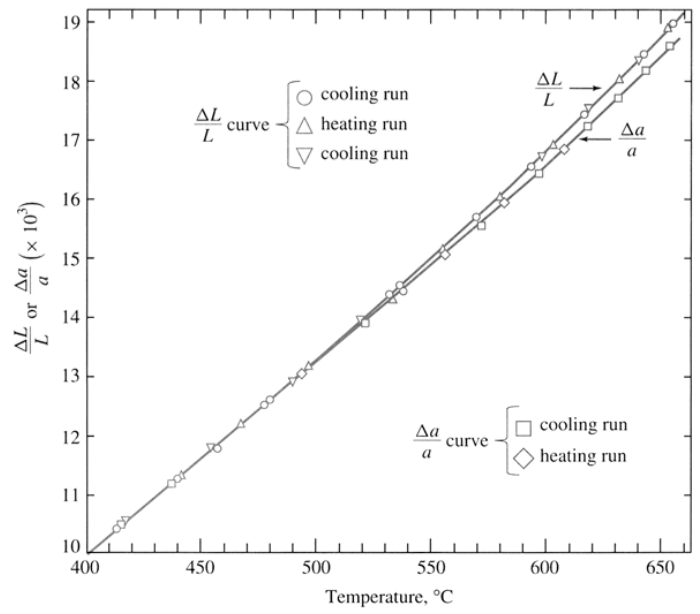


Fig. 4.1 Curves of the thermal expansion of aluminum measured by dilatometry and by x-ray diffraction.